

# Guidelines for Quantifying Plastics in the Marine Environment

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Plastics in the Marine Environment, Trophic System and Aquaculture in the Philippines  
(PlasMics)  
University of the Philippines – Marine Science Institute  
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# Guidelines for Quantifying Plastics in the Marine Environment

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- 1 Introduction
- 1.1 Background
- 1.2 Why the need to monitor plastics in the marine environment

## 2 Sampling methods

## 2.1 Contamination Control

- All sample containers and sampling equipment should be pre-washed with acid and then with filtered distilled water before use. After drying, wrap all metal equipment to be used.
- Avoid using any plastic material, if possible.
- Wear polymer-free clothing or cotton overalls and gloves. The type and color of each person should also be noted.
- Position people down-wind of the sampling apparatus during deployment and recovery.
- Cover sample containers with aluminum foil to prevent sample contamination.
- Use pre-filtered distilled water when washing the equipment in between samples.
- During sediment sampling, place three wet filter papers in glass petri dish in the sampling area to account for contamination from air.
- During water sampling, fill sampling jars with filtered seawater or filtered distilled water to serve as blanks.

## 2.2 Sediments

### 2.2.1 Macro- and Microplastics on/in sediments

(Modified from NOAA [Lippiatt et al. 2013] and the European Commission – Joint Research Centre for the Marine Strategy Framework Directive [2013], Adapted from MicroSEAP)

#### **Introduction**

Plastics are synthetic hydrocarbon polymers used by humans for its durability and wide range of applications (Thompson et al. 2009). We continually produce plastics because of our dependence on the material, but as synthetic polymers with chemical additives, these take hundreds of years to decompose, accumulating in landfills or in natural environments (Geyer et al. 2017). In 2010, Jambeck et al. (2015) estimated that 4.8 to 12.7 million metric tons of plastics was released into the ocean. Plastic wastes remain afloat at sea or sink down the seafloor. However, majority of these wastes were estimated to accumulate in coastal regions (Lebreton et al. 2019), leaving coastal areas and sediments at risk to plastic pollution (Barnes et al. 2009; Pinnell and Turner 2019). Thus, it is important to quantify plastic debris in coastal areas to measure the extent of pollution.

This document outlines methods for macro-, meso-, and micro- assessment in coastal areas. This also provides a comparison between rapid and comprehensive survey methods, helping develop standardized and optimize methods for shoreline surveys in the Philippines. The supplied methods will determine the debris density (# of plastic pieces/unit area) and the type of debris materials.

#### **Site selection**

- Sandy or pebble shoreline
- A minimum length of 100 m parallel to the water
- Low to moderate slope (15 to 45 degrees)
- Clear access to the sea (no breakwater or jetties)
- Accessible to survey teams year round
- Must not be part of a clean-up program

#### **Sampling**

Sampling should be done four times a year, once during the northeast monsoon (Amihan), once during the inter-monsoon from Amihan to southwest monsoon (Habagat), once during the Habagat season, and once during the inter-monsoon from Habagat to Amihan. This is done to look at the possible effects of monsoon on debris loading on shorelines.

Sampling should be started within an hour from the low tide and up to within three hours from the actual low tide to analyze the maximum width of the shoreline.

#### **Materials/Equipment:**

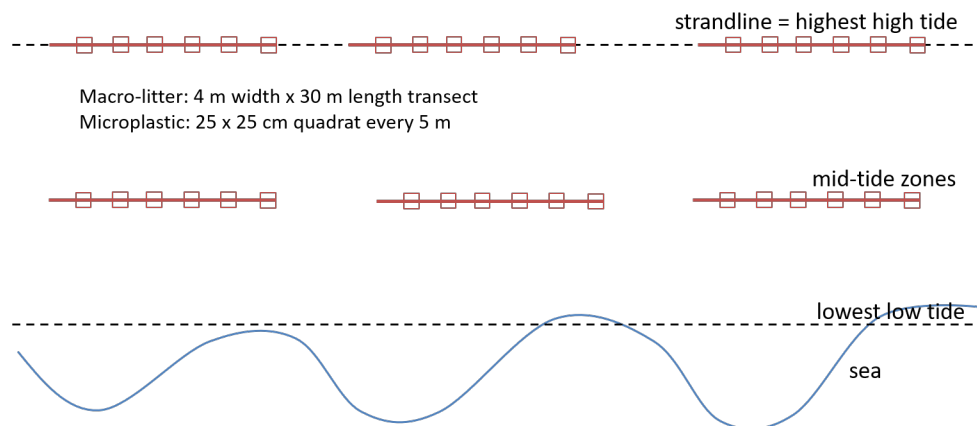
- 30 m transect tapes (x 3)
- 25 × 25 cm metal quadrat
- 8 cm diameter metal ring
- 250-mL glass jars
- Metal shovel / spoon
- Metal ruler
- Sampling bottles to collect water samples for nutrient analysis
- Probes to measure temperature, dissolved oxygen, salinity, etc
- Data sheets, pencils
- Digital camera



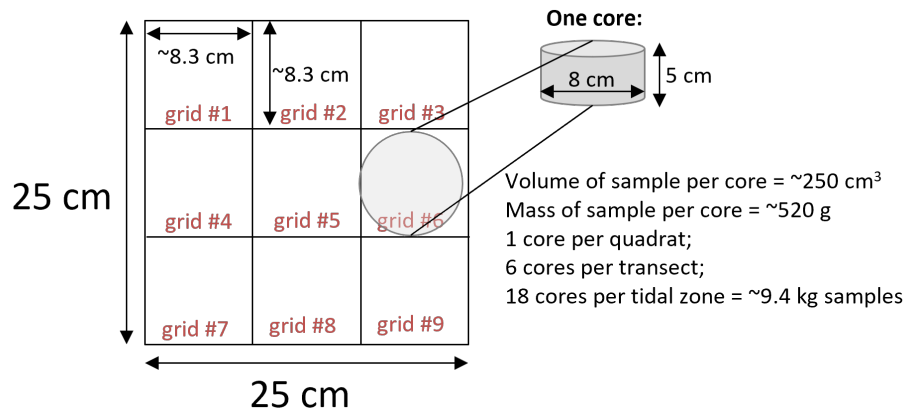
Hand-held GPS unit  
 Extra batteries (rechargeable batteries)  
 Flag markers/stakes  
 10-m measuring tape or surveyor's wheel  
 First aid kit (sunscreen, bug spray, drinking water)  
 Work gloves  
 Clipboard for each surveyor  
 Data sheets (printed on waterproof paper if possible)  
 Pencils for data recording  
 Labels and markers

## Field Survey

1. Depending on the type of habitat, lay the following number of transects:
  - a. For mangrove: lay three 30 m transects in the mid-tide zones and additional three 30 m transects at the strandline.
  - b. For beach: lay three 30 m transects at the mid-tide zones and additional three 30 m transects in the strandline.
  - c. For seagrass: lay three 30 m transects in the mid-tide zones.
  - d. For coral reef: lay three 30 m transects on the reef crest.
  - e. The transects are laid parallel to the shoreline (see illustration below). The minimum distance between two consecutive transects should be at least 2 m apart or wider. For larger study sites, the distance between two consecutive transects can be larger.



- f. For smaller sites that cannot fit 3 x 30 m transects, shorter transect length (e.g., 15 or 20 m) may also be adopted, but the number of transects at each tidal zone should still be three.
2. For sampling macro-litter (> 25 mm in size), record the following information for each litter item found within 2 m to the left and 2 m to the right of the transects (i.e., 4 m total width x 30 m total length belt transect).
3. Place a 25 × 25 cm quadrat every 5 m (i.e., at 5, 10, 15, 20, 25, and 30-m mark). Take a top-down view photo of the entire quadrat (as vertical as possible). Include the entire quadrat frame and tape measure.
4. For sampling mesoplastics (5 – 25 mm) and microplastics (<5 mm):
  - a. Divide each quadrat into 3 × 3 grids, and randomly select one grid (use random number generator to generate one number from 1 to 9) for coring (8 cm diameter, 5 cm depth).



- b. If there is an obstacle (e.g., plant, rock, coral) that prevents quadrat to be placed at the transect mark, scan clockwise at a radius of 5 m looking for the first substratum free to place the quadrat.
  - c. If the surface of the sediment within the core is uneven, use a scraper to flatten the surface. This is to keep sample depth equal for each core.
  - d. Use a scoop and/or scraper to take out sediment samples from the top 5 cm of the surface within the core. Use metal ruler to control the depth.
  - e. Store each core sample in glass jars.
  - f. Field blanks: place three wetted filter papers on metal trays during sediment collection in the field. Store the filter papers in glass petri dishes and bring back to lab for checking under microscope.
5. Collect the following environmental parameters at each site:
    - a. Tidal distance (maximum horizontal distance between the low- and high-tide line) and the distance between strandline and transect.
    - b. Depth of transects (for coral reefs site; using dive watch)
    - c. Wind speed and direction (determined using anemometer / phone app, or simply leeward / windward)
    - d. Three water samples (250 ml for nitrate, phosphate analyses)
    - e. Temperature, dissolved oxygen, salinity, light levels, total suspended solids (using YSI probes or other available equipment; 3 point samplings per site)
    - f. Note: See datasheet template for other site characteristics that can be filled before / after the field survey.
    - g. Note: Other environmental data related to seasons (e.g. average monthly rainfall) will be extracted from nearby meteorological stations.
  6. Store sediment samples at 4 ° C (or lower) for short-term storage (< 2 weeks) or -20 ° C (or lower) for long-term storage until further lab processing.

The macrodebris item concentration (number of debris items/m<sup>2</sup>) per transect is calculated as follows:

$$C = (n)/wl$$

Where  $C$  = concentration of debris items (# of debris items/square meter)  
 $n$  = # of macro-debris items observed  
 $w$  = width (m) of shoreline section recorded during sampling (i.e. transect width)  
 $l$  = length (m) of the shoreline sampled = 30

For a given sampling event

1. The debris concentration for each transect is surveyed (a minimum of four per survey).
2. Take the mean concentration at each transect to calculate an overall site concentration ( $\pm$  standard deviation) for that date.

## 2.3 Water

### 2.3.1 Sea surface

(Adapted from Lippiatt et al. 2013, Green et al. 2018, Michida et al. 2019, GESAMP 2019)

Manta net tows and grab sampling should be done in tandem since net tows alone can underestimate smaller microplastics, those particles smaller than the size of the mesh. Since obtaining grab samples does not initially involve pre-filtering of samples prior to analysis in the laboratory, this sampling technique can collect microplastics smaller than the mesh size of the net.

Manta net tows should be done under calm sea conditions and no algal blooms. Wave height should be under 0.5 meters and Beaufort wind force should be under three (GESAMP 2019). A constant immersion depth should be maintained all throughout the sampling.

#### **Materials/Equipment:**

- Manta net
- Rope for manta
- Weights
- Pre-filtered distilled water
- Flow meter
- Anemometer
- Current meter
- Multiparameter meter
- 1000 mL sampling jars
- 1000 mL HPDE bottles for chlorophyll-a
- 1000 mL HDPE bottles for total suspended solids
- 250 mL HDPE bottles for nutrients
- Labels and markers
- Spray bottle or hose for washing the net
- Glass petri dishes
- GF/F filter papers
- Metal forceps for GF/F filter papers
- Foil for closing sample jars
- Long gloves for sampling
- Cooler with ice

#### **Chemicals:**

- Isopropanol for preservation

#### **Procedure:**

- 1) Before going out to sea, wash manta net from the outside with pre-filtered distilled water (PFDW). Store in a natural fiber bag.
- 2) At sea and before deployment, wash net from the outside using PFDW using a garden sprayer. If there is water on the ship available, wash using the hose. Concentrate the contents to the cod end and label as one of your blanks. Store together with samples.
  - MNLBAY\_04May2021\_Blank\_Manta\_S1
  - [cruise\_date\_Blank\_Manta\_site/station#]
- 3) Set up other blanks: petri dish with PFDW and petri dish with wet GF/F (accomplish in triplicates).
  - MNLBAY\_04May2021\_Blank\_PD\_S1a
  - MNLBAY\_04May2021\_Blank\_GFF\_S1a
  - [cruise\_date\_Blank\_PD or GFF\_site/station#]
  - where PD = petri dish

- GFF = wet filter paper  
 1a = station or site 1, replicate 1 (b and c for second and third replicates)
- 4) Before sampling, fill three 1000-mL sampling jar with 500 mL pre-filtered distilled water. These three will also serve as your blanks.  
 MNLBAY\_04May2021\_Blank\_Grab\_S1a  
 [cruise\_date\_Blank\_Grab\_site/station#]  
 where Grab = blank for grab  
 1a = station or site 1, replicate 1 (b and c for second and third replicates)
  - 5) Attach flowmeter and weights to net.
  - 6) Wash net twice and deploy net at the side of the boat.
  - 7) Collect metadata before starting.
  - 8) Obtain 1000 mL of seawater for grab sample of sub-surface water (~50 cm) and cap while under water to prevent airborne contamination.
  - 9) then seawater for nutrients, chlorophyll-a, total suspended solids, and environmental parameters using a multiparameter meter.
  - 10) Tow for 20 min at 1-2 knots.
  - 11) At the halfway mark or after 10 minutes, obtain again 1000 mL of seawater for grab sample, then seawater for nutrients, chlorophyll-a, total suspended solids, and environmental parameters using a multiparameter meter.
  - 12) After 20 minutes, retrieve and hang net.
  - 13) While allowing the net to hang, obtain 1000 mL of seawater for grab sample, then seawater for nutrients, chlorophyll-a, total suspended solids, and environmental parameters. Also record required metadata.
  - 14) Wash net from the outside using hose or garden sprayer, concentrating contents to cod end.
  - 15) Drain cod end contents to labelled glass jars:  
 MNLBAY\_04May2021\_Manta\_S1  
 [cruise\_date\_Manta\_site/station#]
  - 16) Add isopropanol (10% of the total sample volume).
  - 17) Cover petri dishes and glass jars (with foil before replacing lids) and store with samples.
  - 18) Storage: -20° C to 4° C (may be best to store in 4° C as the glass may break).

### 2.3.2 Water column

#### **Materials/Equipment:**

Plankton net  
 Rope for plankton net  
 Weights  
 Pre-filtered distilled water  
 250 mL or 300 mL glass sampling jars  
 Labels and markers  
 Spray bottle  
 Glass petri dishes  
 GF/F filter papers  
 Metal forceps for GF/F filter papers  
 Foil for closing sample jars  
 Gloves

#### **Chemicals:**

Isopropanol for preservation

#### **Procedure:**

- 1) Before going out to sea, wash plankton net from the outside with pre-filtered distilled water (PFDW). Store in a natural fiber bag.

- 2) At sea and before deployment, wash net from the outside using PFDW using a garden sprayer. If there is water on the ship available, wash using the hose. Concentrate the contents to the cod end and label as one of your blanks. Store together with samples.  
WPS\_04May2021\_Blank\_net\_S1  
[cruise\_date\_Blank\_net\_site/station#]
- 3) Set up other blanks: petri dish with PFDW and petri dish with wet GF/F (accomplish in triplicates).  
WPS\_04May2021\_Blank\_PD\_S1a  
WPS\_04May2021\_Blank\_GFF\_S1a  
[cruise\_date\_Blank\_PD or GFF\_site/station#]  
where PD = petri dish  
GFF = wet filter paper  
1a = station or site 1, replicate 1 (b and c for second and third replicates)
- 4) Attach weights to net.
- 5) Wash net twice and deploy net at the surface (or whichever depth is preferred).
- 6) Retrieve net and hang, allow contents to concentrate to cod end.
- 7) Wash net from the outside using hose or garden sprayer, concentrating contents to cod end.
- 8) Drain cod end contents to labelled glass jars:  
WPS\_04May2021\_Net\_S1  
[cruise\_date\_Net\_site/station#]
- 9) Add isopropanol (10% of the total sample volume).
- 10) Cover petri dishes and store with samples.
- 11) Storage: -20° C to 4° C (may be best to store in 4° C as the glass may break).

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Note: Seawater samples from the water column should have accompanying metadata.

## 2.4 Biota

### **Introduction**

Plastic debris as emerging pollutants in the environment are found to degrade into microplastics which are characterized as fibers, fragments, and pellets that are less than 5 mm in size (Andrady et al. 2011). The ubiquity of microplastics, which have been detected in a variety of habitats, has become a threat to marine species, ecosystems, and human health. The ingestion of microplastics by marine biota across different trophic levels, including organisms significant in mariculture, has been reported by various studies (Lushar et al. 2015; Argamino and Janairo 2016; Karlsson et al. 2017; Naidoo et al. 2020). Microplastics ingestion was found to cause internal blockages and injury to the digestive tract of organisms, and has been speculated as the initial route for associated toxic chemicals to enter and spread through the food web with microplastics as vectors, culminating in human consumption (Nadal et al. 2016; Diepens and Koelmanns 2018). However, several processes that contribute to possible biomagnification of species remain little understood. Therefore, quantification of ingested and accumulated microplastics in biota is a significant step in understanding such processes and assessing the risk of such hazards, as a baseline data for setting rules and guidelines for the trade and consumption of mariculture species.

This document provides methods for quantification of microplastics in biota, specifically in bivalves (mussels and oysters) and fish. The methods listed will outline protocols from sampling, quality controls, sample processing, until microplastics identification, characterization, and quantification. The protocols will determine microplastics type and abundance as number of plastic particles per gram of wet weight.

### **Materials and Equipment**

- Aluminum foil
- Ice box
- Ziplock bags
- Rubber bands
- Dissecting pans

### **Reagents**

- Sterile filtered distilled water

### **Procedure:**

1. Collect 50 individuals from each sampling site. Wash the samples with pre-filtered water then wrap with aluminum foil and rubber bands to prevent contamination. Store in an ice box and transfer to a -20° C freezer until lab processing.

### 3 Laboratory Analyses

### 3.1 Contamination Control

- All materials, equipment, and laboratory surfaces to be used should be thoroughly rinsed and cleaned with pre-filtered water and stored under clean air conditions.
- Reagents and solutions should be pre-filtered before treatment.
- Sample handling and processing should be conducted in clean air facilities. If clean air conditions cannot be fully achieved, use of negative controls needs to be established.
- Triplicate negative controls need to be included in the treatments for each batch of sample processed.
- For additional controls, clean Petri dishes or glass slides should be placed around the area for sample processing to check for airborne contamination.
- Samples should be covered with aluminum foil before and after processing. Polymer-free gloves and laboratory coat made of 100% cotton material should be worn during analysis to prevent contamination.



## 3.2 Preparation of spiked samples

### 3.2.1 Preparation of microplastics

(Methods from Nuelle et al. 2014; Kuhn et al. 2018; Selonen et al. 2020)

#### **Materials/Equipment:**

Grinder or mill  
Liquid nitrogen  
Sieves of varying sizes  
Fleece blanket – brightly-colored  
Dog brush  
Cutter  
Cutting mat  
Plastics:  
    PET – water bottle  
    PP – straw  
    LDPE – ice bag  
    HDPE – shampoo container  
    PVC – pipe  
    PS – yogurt cup  
Fibers:  
    Nylon, PET, and PP

#### **Procedure:**

1. Cut plastics into smaller pieces, less than 25mm. Soak in liquid nitrogen for 30 minutes to increase brittleness. Grind or mill plastic pieces. Sieve plastic pieces through 5mm, 1mm, and 100  $\mu$  m. Sort plastics per size and per type and store in glass containers.
2. Plastics can also be cut using a cutter to produce larger MPs (1mm to 5mm). Sieve cut plastics through through 5mm, 1mm, and 100  $\mu$  m, sort per size and per type and store in glass containers.
3. To obtain long fibers, brush a fleece blanket using a dog brush with bent wires. Store in glass containers. For short fibers, cut fleece blanket into small pieces, 0.5 square centimeter. Soak cut blanket in liquid nitrogen for 5 min before milling for 2.5 minutes. Store milled fibers in a sealed paper container or glass container.

### 3.2.2 Organic matter

(Methods from Isobe et al. 2019)

#### **Materials/Equipment:**

Glass containers  
Wood chips  
Bivalve shells  
Crab shells  
Eggshells  
Zooplankton (optional)  
Grinder  
Sieves (5mm, 1mm, and 0.3mm)

#### **Procedure:**

4. Mechanically fragment organic matter using a grinder. Sieve fragmented organic matter through 5mm, 1mm, and 0.3mm sieves, sort according to type and size, and store in glass containers.

### 3.2.3 Sediments

(Modified from Shim et al. 2016)

#### **Materials/Equipment:**

Beach sand  
Sieves (1000  $\mu\text{m}$ , 63  $\mu\text{m}$ )  
Density separation set-up  
Furnace  
Furnace-friendly container for sand combustion  
Prepared microplastics (MPs) for spiking

#### **Chemicals:**

Zinc chloride for density separation

#### **Procedure:**

1. Collect natural sand from a beach.
2. Remove light particles and plastics by density separation. Repeat until no more particles are visible.
3. Pre-combust cleaned sand overnight at 450 ° C to remove the remaining plastics and organic matter.
4. Add known amount of MPs to spike 50 g of the pre-combusted sand in a beaker. Also add organic matter into the mixture. Mix.
5. Analyze this sand and plastic mixture along with samples to test for extraction efficiency.

### 3.2.4 Seawater (Modified from Isobe et al. 2019)

#### **Materials/Equipment:**

Seawater  
Filters: 10  $\mu$  m and 1  $\mu$  m  
1 L glass bottle container  
Prepared microplastics (MPs) for spiking

#### **Procedure:**

1. Filter seawater with 10  $\mu$  m filter twice and with 1  $\mu$  m filter before filling sample bottles. Make sure there are no plastic fragments in the water and in the sample bottles.
2. Add known amount of MPs per type and size to the sample bottle containing filtered seawater.
3. Add organic material to the sample bottle. Mix.
4. Analyze this mixture along with samples to test for extraction efficiency.

### 3.3 Extraction of MPs from environmental samples

#### 3.3.1 Sediments

##### **Materials/Equipment:**

Stainless steel sieves, each measuring 8 in (diameter) and 2 in (depth)

- o 5.6 mm mesh (Number 3.5)
- o 1 mm mesh (Number 18)
- o 0.3 mm mesh (Number 50)
- o 0.037 mm mesh (Number 400) – or smallest limit of detection

Squirt bottle containing distilled water  
500-mL glass beaker  
Analytical balance (precise to 0.1 mg)  
Metal spatula  
Drying oven (90°C)  
Stir bar  
Laboratory hot plate  
Watch glass  
Standard Metal Forceps  
Density separator, which is assembled using the following method:

- o A glass funnel (122-mm in diameter) is fitted with a 50-mm segment of latex tubing on the bottom of the stem and a pinch clamp is attached to control liquid flow from the funnel.

Retort stand  
O-ring  
Spring clamp (2-inch)  
Aluminum foil  
4-mL glass vials  
Dissecting microscope (40X magnification)

##### **Reagents:**

ZnCl<sub>2</sub> salt solution with density of 1.5 g cm<sup>-3</sup>

- a) Mix 972 g of ZnCl<sub>2</sub> in 1 l of DI / Milli-Q water, stirred for at least 24 h at 600 rpm. Adjust the density to 1.5 g cm<sup>-3</sup> by checking the mass and volume of the solution.
- b) If white precipitate is observed, add concentrated HCl until the solution turns clear (pH of solution should be <5)
- c) Filter using 0.45 µm filter papers to remove impurities.

Iron (Fe(II)) solution (0.05 M)

- a) Prepared by adding 7.5 g of FeSO<sub>4</sub> · 7H<sub>2</sub>O (= 278.02 g/mol) to 500 mL of water and 3mL of concentrated sulfuric acid
- b) Filter using 0.45 µm filter papers to remove impurities.

30% Hydrogen peroxide

##### **Procedure:**

1. Dry sediment at 50 ° C for at least 48 h or until sample dryness. Record the dry mass of the sediment sample.
2. Sieve the sediment sample through a 5 mm metal sieve (and any other smaller sieve sizes if you are looking at sediment size grains) on top of a metal tray. Record the dry mass of sediment for each size category (< 5 mm, > 5 mm).
3. For sediment samples > 5 mm, visually select plastic-like particles using forceps and record the number and mass.

4. For sediment samples < 5 mm, conduct density-separation method:
  - a. Prepare  $\text{ZnCl}_2$  salt solution
  - b. Add salt solution at a ratio of 100 g dried sediment to 400 ml salt solution.
  - c. Mix sediment and salt solution thoroughly for 2 min. Remove the magnet from the mixture solution.
  - d. Cover the beaker with aluminium foil and let the particles settle >8 h or overnight.
  - e. Gently pour only the supernatant from the beaker and filter through 20  $\mu\text{m}$  under vacuum.
    - i. Filter 5% of the resultant filtrate through 0.22  $\mu\text{m}$  filter for analysis by microRaman (to be sent to the UK team).
    - ii. Collect used salt solution and filter them through 0.22  $\mu\text{m}$  filter to recycle for future use (up to 5x).
  - f. Rinse the filter paper with enough amount of DI / Milli-Q water to remove the salt solution
  - g. Transfer the particles on filter paper into a 250 ml beaker with as little DI / Milli-Q water as possible. Cover the glass beaker with aluminium foil, but one edge is slightly opened.
  - h. Dry the beaker at 50° C in the oven for 24 h or longer until sample dryness.
5. Conduct oxidative digestion:
  - a. Prepare 0.05M Fe (II) solution.
  - b. Add magnetic stirrer to the 250 ml beaker (with microplastics obtained from Step i in density-separation method).
  - c. Add 20 ml of Fe(II) solution and 20 ml of 30%  $\text{H}_2\text{O}_2$  solution into the beaker.
  - d. Stir the solution at room temperature.
    - i. Monitor the temperature and keep the solution between 20–40° C. Put solution in ice bath intermittently when temperature is >40° C. Temperature should not fall <15° C as yellow precipitate will form. If yellow precipitate is formed, add concentrated  $\text{H}_2\text{SO}_4$  to dissolve it.
  - e. If the solution is still brownish color, add more Fe(II) solution and  $\text{H}_2\text{O}_2$  (repeat step c and d).
  - f. Weigh clean filter paper.
  - g. Let the solution cool down. Gently pour the solution over the filter paper.
  - h. Rinse the filter paper with enough amount of DI / Milli-Q water to remove any chemical.
  - i. Transfer the filter paper into the glass petri dish / crucibles. Cover the sample container with aluminium foil, but one edge is slightly opened.
  - j. Dry samples at 50° C in an oven for 1–3 days (until no change in mass).
  - k. Weigh the filter paper.
6. Repeat density-separation and/or digestion steps if filter paper in Step 5k is still dirty.
7. Store filter papers in sealed glass containers until further processing.

### 3.3.2 Water

#### Materials/Equipment:

- Stainless steel sieves, each measuring 8 in (diameter) and 2 in (depth)
  - o 5.6 mm mesh (Number 3.5)
  - o 1 mm mesh (Number 18)
  - o 0.3 mm mesh (Number 50)
  - o 0.037 mm mesh (Number 400) – or smallest limit of detection
- Squirt bottle containing distilled water
- 250-mL glass beaker
- Analytical balance (precise to 0.1 mg)
- Metal spatula
- Drying oven (90°C)
- Stir bar
- Laboratory hot plate
- Watch glass
- Standard Metal Forceps
- Density separator, which is assembled using the following method:
  - o A glass funnel (122-mm in diameter) is fitted with a 50-mm segment of latex tubing on the bottom of the stem and a pinch clamp is attached to control liquid flow from the funnel.
- Retort stand
- O-ring
- Spring clamp (2-inch)
- Aluminum foil
- 4-mL glass vials
- Dissecting microscope (40X magnification)

#### Reagents:

- ZnCl<sub>2</sub> salt solution with density of 1.5 g cm<sup>-3</sup>
  - d) Mix 972 g of ZnCl<sub>2</sub> in 1 l of DI / Milli-Q water, stirred for at least 24 h at 600 rpm. Adjust the density to 1.5 g cm<sup>-3</sup> by checking the mass and volume of the solution.
  - e) If white precipitate is observed, add concentrated HCl until the solution turns clear (pH of solution should be <5)
  - f) Filter using 0.45 µm filter papers to remove impurities.
- Iron (Fe(II)) solution (0.05 M)
  - c) Prepared by adding 7.5 g of FeSO<sub>4</sub> · 7H<sub>2</sub>O (= 278.02 g/mol) to 500 mL of water and 3mL of concentrated sulfuric acid
  - d) Filter using 0.45 µm filter papers to remove impurities.
- 30% Hydrogen peroxide

#### Procedure:

1. Filter water samples through 20 µm filter (if it is not done in the field).
  - a. Transfer the particles on 20 µm filter into a 250 ml beaker with as little DI / Milli-Q water as possible. Cover the glass beaker with aluminium foil, but one edge is slightly opened.
  - b. Filter 5% of the resultant filtrate through 0.22 µm filter for analysis by microRaman (to be sent to the UK team).
2. Dry the beaker at 50° C in the oven for 24 h or longer until sample dryness.
3. If the water samples are clear of sediment and contain mainly planktons, conduct oxidative digestion first followed by density-separation.

If the water samples are turbid and contain mainly fine silt, conduct density-separation first followed by oxidative digestion.

4. Oxidative digestion:
  - a. Prepare 0.05 M Fe(II) solution:
  - b. Add magnetic stirrer to the 250 ml beaker (from step 2 above).
  - c. Add 20 ml of Fe(II) solution and 20 ml of 30% H<sub>2</sub>O<sub>2</sub> solution into the beaker.
  - d. Stir the solution at room temperature.
    - i. Monitor the temperature and keep the solution between 20–40 ° C. Put solution in ice bath intermittently when temperature is >40 ° C. Temperature should not fall <15 ° C as yellow precipitate will form. If yellow precipitate is formed, add concentrated H<sub>2</sub>SO<sub>4</sub> to dissolve it.
  - e. If the solution is still brownish color, add another Fe(II) solution and H<sub>2</sub>O<sub>2</sub> (repeat step c and d).
  - f. Weigh clean filter paper.
  - g. Let the solution cool down. Gently pour the solution over the filter paper.
  - h. Rinse the filter paper with enough amount of DI / Milli-Q water to remove any chemical.
5. Density-separation method:
  - a. Prepare ZnCl<sub>2</sub> salt solution with density of 1.5 g cm<sup>-3</sup>
  - b. Transfer the particles found on filter paper from step 4h into a beaker using the ZnCl<sub>2</sub> salt solution. Add 400 ml salt solution for every 100 g of particles.
  - c. Mix sediment and salt solution thoroughly for 2 min. Remove the magnet from the mixture solution.
  - d. Cover the beaker with aluminium foil and leave the beaker overnight
  - e. Weigh clean filter paper.
  - f. Gently pour the supernatant from the beaker and filter through filter paper.
    - i. Collect used salt solution and filter them through 0.22 um filter to recycle for future use (up to 5x?).
  - g. Rinse the filter paper with enough amount of DI / Milli-Q water to remove the salt solution.
6. Transfer the particles on filter paper (from Step 4h or 5g) into a glass petri dish / crucible with as little DI / Milli-Q water as possible. Cover the sample container with aluminium foil, but one edge is slightly opened.
7. Dry samples at 50° C in an oven for 1–3 days (until no change in mass).
8. Weigh the filter paper.

### 3.3.4 Biota

#### Materials and Equipment

Aluminum foil  
Ice box  
Ziplock bags  
Rubber bands  
Dissecting kit (scalpel, forceps, needle, scissors)  
Dissecting pans  
Stainless steel caliper  
Stainless steel shucking knife  
Metal spatula  
Beakers (250 ml, 600 ml)  
Petri dishes (60 mm, 100 mm)  
Glass microfiber filters (1  $\mu$ m, 47 mm)  
Buchner funnel  
Buchner flask  
Stopper and rubber tubing  
Vacuum pump  
Stereomicroscope  
Blue LED lamp (450 nm)  
Top loading balance

#### Reagents

Potassium hydroxide  
Hydrogen peroxide  
Sodium chloride  
Nile red  
Sterile filtered distilled water

#### Procedure

##### A. Bivalves

1. Using a stainless-steel shucking knife, open the shells and empty the contents of 6 individuals into a clean 1L beaker.
2. Add 10% KOH 4 times more than the tissue volume, cover the beaker opening with foil and glass cover, and stand at 60° C for 48 hours. Agitate the set-up manually every 4 hours.
3. Group the beakers into 2 set-ups: Set-up A and Set-up B. Set-up A will follow the protocol established by Bendell et al. 2020, while Set-up B will proceed according to the methods conducted by Li et al. 2015.
4. For Set-up A, add 2.5 mL of Dye More, and vacuum filter the digested solution through #1 GF/F filter paper. Wash the filter paper with 30% H<sub>2</sub>O<sub>2</sub>, and afterwards stain the filter paper with Nile red.
5. For Set-up B, add 800 mL saturated NaCl into the digested solution and mix. Allow to settle overnight. Repeat this step for 3 more times, then vacuum filter through 1.0  $\mu$ m GF/F. Stain the filter paper with Nile red.
6. Store the filter paper in a sterile petri dish, with the lid wrapped by foil to prevent contamination, until MP quantification.

##### B. Fish

1. Rinse the exterior of the fish thoroughly before placing in the dissection area to reduce particle contamination during processing.
2. Using a sterile scalpel blade, make an incision 1 mm in front of the rectum and cut towards the anterior part of the fish.



3. In order to estimate ingested microplastics accurately, the entire gastrointestinal tract should be examined. Cut through the esophagus while keeping the entire stomach intact, and cut through the gut approximately 2-3 mm before the anus. Record the weight of the stomach and gut.
4. Place the excised stomach and gut in a clean 1L beaker. Add 10% KOH solution 3 times more than the tissue volume. Cover the beaker opening with foil and glass cover, and stand at 60° C for 48 hours. Agitate the set-up manually every 4 hours.
5. After digestion, add 800 mL saturated NaCl to separate plastics from digested tissue via floatation. Allow the solution to stand overnight until clearance is observed. Filter solution through 1.0 um GF/F filter.
6. Stain the filter paper with Nile red. Place filter paper in a sterile petri dish then wrap with foil for storage until MP quantification.

### 3.4 Counting, visual identification, and sorting

#### **Materials and Equipment**

Stereomicroscope with camera  
Fluorescent blue lamp  
Computer  
Extracted microplastics on filter papers

#### **Reagents**

Ethanol

#### **Procedure**

1. Subject MPs in filter papers to initial visual inspection and quantification under a stereomicroscope. Observe stained filter paper under fluorescent blue lamp (420-495 nm) to detect fluorescing MPs.
2. Characterize fluorescent MPs according to shape (fibers, spherules, fragments, or sheets), colors, and sizes. Record colors as dominant visual property and categorize sizes into five groups: 0.05-0.1 mm, 0.1-0.5 mm, 0.5-1 mm, 1-1.5 mm, 1.5-2 mm.
3. The abundance of microplastics in the samples will be calculated as the number of plastic particles per gram of wet weight.

### 3.5 Characterization

(Adapted from Harshvardhan and Jha (2013), Jung et al. (2018))

To check for the formation of new functional groups, which can be attributed to polymer degradation, Fourier Transform Infrared Spectroscopy with Attenuated Total Reflection (ATR) (Bruker, Massachusetts, United States) is done.

#### **Materials and Equipment**

Oven  
Fourier Transform infrared spectrophotometer with attenuated total reflectance  
Computer  
Forceps

#### **Reagents**

70% Ethanol

#### **Procedure**

1. Among the identified MPs, random samples will be selected from each size category and will be further verified through FT-IR spectroscopy.
2. Dry samples overnight in oven at 60° C.
3. Collect spectra from plastic samples from 4,000  $\text{cm}^{-1}$  to 450  $\text{cm}^{-1}$  with a data interval of 1  $\text{cm}^{-1}$ , with the resolution set to 2  $\text{cm}^{-1}$ .
4. Clean the ATR diamond with 70% ethanol before and in-between samples.
5. Perform background scans between samples.
6. Each sample is compressed against the diamond with the minimum force recommended by the manufacturer to ensure good contact between the sample and the ATR crystal.
7. The resultant spectra will be identified using the spectra library in the software and will be use to determine polymer types.

## 4 References

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## 5 Appendices

## MACRO- AND MICROPLASTIC DATASHEET – SITE CHARACTERISTICS

Surveyor name:				Survey date:			
Site name:				Country:			
GPS start:				GPS end:			
<b>SHORELINE CHARACTERISTICS</b>							
Habitat type: beach mangrove seagrass coral reef							
Backshore type: cliff seawall urban building forest/tree shrub mangrove other:							
Shore exposure: cove/bay straight headland							
Tidal distance (m):							
<b>ENVIRONMENTAL CONDITIONS (three readings per site)</b>							
Weather: clear rain/storm overcast drizzle							
Wind speed (m/s):				Wind direction:			
Sea surface temp (°C):				Salinity (psu):			
Dissolved O <sub>2</sub> (mg/L):				Total suspended solids (mg/L):			
Nitrate (mg/L):				Phosphate (mg/L):			
<b>LAND-USE CHARACTERISTICS</b>							
Access: vehicular trail isolated							
Major site usage: tourism fishing protected isolated other:							
Nearest town distance (km):				Nearest river distance (km):			
River input: yes no				Pipe/drain input: yes no			
Evidence of dumping: none construction household other:							
Evidence of recent activities: none clean-up/rubbish removal apparent spilled trash storm/flood strong winds public event other:							
Notes ( <i>include descriptions on landmarks, coastal hydrography, etc</i> ):							

**Notes:**

GPS start: GPS coordinates of the start of the first transect

GPS end: GPS coordinates of the end of the third transect

Tidal distance: the maximum horizontal distance between the low- and high-tide line

Access: vehicular (you can drive to the site), trail (you must walk), isolated (you need a boat/plane)

Nearest town distance: nearest distance to villages / residences / towns / human populations

## MACRO- AND MICROPLASTIC DATASHEET

Surveyor name:		Survey date:		
Site name:		Country:		
Transect number:     out of		Transect size (length × width):		
Distance from strandline (m):		Depth (m):		
Sampling zone:   strandline   mid-tide   reef crest				
Substratum type:    sand   mud   boulder   seagrass   seaweed   coral   other:				
Main category	Specific litter category	Count	Mass (g)	Note*
<b>Plastic</b>	Beverage bottles < 1 L			
	Beverage bottles ≥ 1 L			
	Buckets / jerry cans / drums			
	Caps / lids / covers			
	6-packs rings / drink package rings			
	Straws / pipettes			
	Clear cups / bowls / food containers			
	Foamed cups / bowls / food containers			
	Knives / forks / spoons			
	Bags			
	Thin plastic wraps / labels / packagings			
	Thick plastic wraps / sacks			
	Lighters / matches			
	Cigarette tips / butts / filters			
	Ropes / strings / strapping bands			
	Pipes / hoses			
	Fishing lines / nets / rods			
	Buoys / floats			
	Shampoos / shower gels / toothbrushes			
	Fragments (hard plastic)			
	Fragments (soft plastic / films / sheets)			
	Fragments (foamed)			
Other:				
<b>Rubber</b>	Slippers / flip-flops / shoes / gloves			
	Tires			
	Balloons, balls			
	Rubber bands			
	Other:			
<b>Metal</b>	Aluminium / tin / aerosol cans			
	Bottle caps			
	Buckets / drums			
	Nails / irons			
	Fishing related (lures, hooks, sinkers)			
	Other:			

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Notes:

Distance from strandline: the horizontal distance between the strandline and and transects

Depth: Depth of transects; for coral reef sites only

If the macro-debris cannot be weighed because it is too big / heavily encrusted / soaked, record its size (length, width, height) for mass estimation.



Main category	Specific litter category	Count	Mass (g)	Note*
<b>Glass</b>	Bottles / jars			
	Light bulbs / tubes / globes			
	Fragments			
	Other:			
<b>Wood</b>	Cigarette packs			
	Lighters / matches			
	Paper / newspaper / pieces of papers			
	Crates / boxes / cardboards			
	Fishing traps / pots			
	Ice cream sticks / chopsticks / toothpicks			
	Fragments			
	Other:			
<b>Cloth</b>	Clothes / towels / rags			
	Sacking / gunny sacks / canvas			
	Diapers / sanitary pads			
	Fabric pieces			
	Other:			
<b>Other</b>	Batteries			
	Appliances, electronics			
	Furniture			
	Contraceptions / condoms			
	Masks / gloves / face shields			
	Syringes			

**\*Note:**

If the macro-debris cannot be weighed because it is too big / heavily encrusted / soaked, record its size (length, width, height) for mass estimation.



